


## MATERIAL SAFETY DATA SHEET

### Section 1. Identification

Product Name:	<b>Ammonium Hydroxide</b>
Synonyms:	Ammonium Hydroxide Solutions, Aqua Ammonia, Aqua Ammonia Solutions, Ammonia Solutions, Ammonia Aqueous, Ammonia Water
CAS REGISTRY NO:	1336-21-6
Supplier:	Hyderabad Ammonia & Chemicals Pvt Ltd Sy.No.296/7/4, IDA Bollaram, Medak Dist
Email:	<a href="mailto:info@hyd ammonia.in">info@hyd ammonia.in</a>
Telephone (General):	27752497
<b>Emergency Telephone Number:</b>	<b>9849747405</b>
Recommended Use:	Various Industrial

### Section 2. Hazard(s) Identification

Hazard:	Acute Toxicity, Corrosive, Acute Aquatic Toxicity	
Classification:	Acute Toxicity, Inhalation (Category 4) Acute Toxicity, Oral (Category 4) Skin Corrosion (Category 1B) Serious Eye Damage / Irritation (Category 1) Acute Aquatic Toxicity (Category 1)	Note: (1 - Most Severe / 4 - Least Severe)
Pictogram:		
Signal word:	<b>Danger</b>	
Hazard statements:	Harmful if inhaled. Harmful if swallowed. Causes severe skin burns and eye damage. Very toxic to aquatic life.	
Precautionary statements:	Avoid breathing mist/vapors/spray. Use only outdoors or in a well-ventilated area. Do not eat, drink or smoke when using this product. Wear protective gloves, protective clothing, eye protection, face protection.	

Precautionary statements  
(continued):

IF INHALED: Remove person to fresh air and keep comfortable for breathing. Call a doctor/physician and seek medical attention for severe exposure or if symptoms persist. Specific treatment, see supplemental first aid instructions in Section 4 (First Aid Measures).  
IF SWALLOWED: Rinse mouth. Do NOT induce vomiting. Call a poison center/doctor/physician. See supplemental first aid instructions in Section 4 (First Aid Measures).  
IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water or shower (minimum of 20 minutes). See supplemental first aid instructions in Section 4 (First Aid Measures).  
IF IN EYES: Immediately call a doctor/physician and seek medical attention. Rinse continuously with water for several minutes (minimum of 20 minutes). Specific treatment, see supplemental first aid instructions in Section 4 (First Aid Measures).  
Wash contaminated clothing before reuse.  
Store in a well-ventilated place. Keep container tightly closed. Store locked up.  
Dispose of contents/container in accordance with local, regional, national regulations as applicable. See section 13 (Disposal Considerations).

**NFPA Rating:**



**HMIS Classification:**

AMMONIUM HYDROXIDE	
HEALTH	3
FLAMMABILITY	1
REACTIVITY	0
PERSONAL PROTECTION	H

**NFPA Numbering System:**

0 = Least Hazardous / 4 = Most Hazardous

**HMIS Hazard Index:**

0 = Minimal, 1 = Slight, 2 = Moderate, 3 = Serious, 4 = Severe

**Section 3. Composition / Information on Ingredients**

**CHEMICAL NAME:** Ammonium Hydroxide (Ammonium Hydroxide Solutions 10% to 30%)

**CAS REGISTRY NO:** 1336-21-6

**SYNONYMS:** Ammonium Hydroxide Solutions, Aqua Ammonia, Aqua Ammonia Solutions, Ammonia Solutions, Ammonia Aqueous, Ammonia Water.

**CHEMICAL FAMILY:** Inorganic nitrogen compounds.

**COMPOSITION:** Solutions: Anhydrous Ammonia (10% to 30%); Water (90% to 70%); Density: 16° Baume to 26° Baume.

**Ammonia, Anhydrous: CAS # 7664-41-7; Water: CAS# 7732-18-5**

**Section 4. First Aid Measures**

IF INHALED: Immediately remove person to fresh air and keep comfortable for breathing. In case of severe exposure or if irritation persists, breathing difficulties or respiratory symptoms arise, seek medical attention. If not breathing, administer artificial respiration. If trained to do so, administer supplemental oxygen, if required.

IF ON SKIN (or hair): Immediately take off all contaminated clothing. Flush skin with copious amounts of tepid water for a minimum of 20 minutes. Do not rub or apply topical, occlusive compounds, such as ointments, certain creams, etc., on affected area. For severe exposure or if irritation persists, seek medical attention. Wash contaminated clothing before reuse.

IF IN EYES: Immediately rinse continuously with copious amounts of tepid water for a minimum of 20 minutes. Eyelids should be held apart and away from eyeball for thorough rinsing. Seek medical attention.

IF SWALLOWED: Rinse mouth. Do not induce vomiting. If conscious, give large amounts of water to drink. May drink orange juice, citrus juice or diluted vinegar (1:4) to counteract ammonia. If unconscious, do not give anything by mouth. Seek medical attention.

**NOTE TO PHYSICIAN:** Respiratory injury may appear as a delayed phenomenon. Pulmonary edema may follow chemical bronchitis. Supportive treatment with necessary ventilation actions, including oxygen, may warrant consideration.

## **Section 5. Fire Fighting Measures**

### **EXTINGUISHING MEDIA:**

Water Spray, Water Fog for escaping ammonia gas.

### **SPECIAL FIRE FIGHTING PROCEDURES:**

Must wear protective clothing and a positive pressure SCBA.

Stop flow of liquid if possible.

Use water spray to keep fire-exposed containers cool.

If a portable container (such as a drum, Intermediate Bulk Container [IBC] or trailer) can be moved from the fire area without risk to the individual, do so to prevent the pressure relief valve from discharging or the container from failing.

Stay upwind when containers are threatened.

### **UNUSUAL FIRE AND EXPLOSION HAZARDS:**

When heated, product will give off ammonia vapor, which is a strong irritant to the eye, skin and respiratory tract.

Outdoors, ammonia is not generally a fire hazard. Indoors, in confined areas, ammonia vapors may be a fire hazard, especially if oil or other combustible materials are present.

Combustion may form toxic nitrogen oxides (NO<sub>x</sub>).

## **Section 6. Accidental Release Measures**

### **GENERAL:**

Only properly trained and equipped persons should respond to an ammonium hydroxide release.

Wear eye, hand and respiratory protection and protective clothing; see Section 8, Exposure Controls / Personal Protection.

Stop source of leak if possible, provided it can be done in a safe manner.

Leave the area of a spill by moving laterally and upwind.

Isolate the affected area. Non-responders should evacuate the area, or shelter in place.

### **SPECIFIC STEPS TO BE TAKEN:**

For a hazardous material release response, Level A and/or Level B ensemble including positive-pressure SCBA should be used. A positive pressure SCBA is required for entry into ammonia atmospheres at or above 300 ppm (IDLH).

Stay upwind and use water spray downwind of container to absorb the evolved gas.

Contain spill and runoff from entering drains, sewers, streams, lakes and water systems by utilizing methods such as diking, containment, and absorption.

## **Section 7. Handling and Storage**

### **SPECIAL PRECAUTIONS:**

Only trained persons should handle ammonium hydroxide.

Store in cool, dry and well-ventilated areas, with containers tightly closed.

Keep out of direct sunlight and away from heat sources.

Do not use any non-ferrous metals such as copper, brass, bronze, aluminum, tin, zinc or galvanized metals.

Protect containers from physical damage.

Closed storage tanks should be provided with safety relief valves and vacuum breakers as necessary.

### **VENTILATION:**

Local exhaust should be sufficient to keep ammonia vapor below applicable exposure standards.

### **WORKPLACE PROTECTIVE EQUIPMENT:**

Protective equipment should be stored near, but outside of ammonium hydroxide area. Water for first aid, such as an eyewash station and safety shower should be kept available in the immediate vicinity.

## Section 8. Exposure Controls / Personal Protection

### EXPOSURE LIMITS FOR AMMONIA: (Vapor)

OSHA	50 ppm,	35 mg / m <sup>3</sup> PEL	8 hour TWA
NIOSH	35 ppm,	27 mg / m <sup>3</sup> STEL	15 minutes
	25 ppm,	18 mg / m <sup>3</sup> REL	10 hour TWA
	300 ppm,	IDLH	
ACGIH	25 ppm,	18 mg / m <sup>3</sup> TLV	8 hour TWA
	35 ppm,	27 mg / m <sup>3</sup> STEL	15 minutes

### PROTECTIVE EQUIPMENT:

**EYE/FACE PROTECTION:** Chemical splash goggles should be worn when handling ammonium hydroxide (aqua ammonia). A face shield can be worn over chemical splash goggles as additional protection. Do not wear contact lenses when handling ammonium hydroxide. Refer to 29 CFR 1910.133 for OSHA eye protection requirements.

**SKIN PROTECTION:** Ammonia impervious gloves and clothing (such as neoprene, butyl and Teflon) should be worn to prevent contact during normal operations, such as loading/unloading, transfers, and handling small spills. Chemical boots can be worn as additional protection.

**RESPIRATORY PROTECTION:** Respiratory protection approved by NIOSH for ammonia must be used when applicable safety and health exposure limits are exceeded. For escape in emergencies, NIOSH approved respiratory protection should be used, such as a full-face gas mask and canisters/cartridges approved for ammonia or SCBA. A positive pressure SCBA is required for entry into ammonia atmospheres at or above 300 ppm (IDLH).

Refer to 29 CFR 1910.134 and ANSI: Z88.2 for OSHA respiratory protection requirements.

**VENTILATION:** Local exhaust should be sufficient to keep ammonia vapor below applicable exposure standards.

**FOR A HAZARDOUS MATERIAL RELEASE RESPONSE:** Level A and/or Level B ensemble including positive-pressure SCBA should be used. A positive pressure SCBA is required for entry into ammonia atmospheres at or above 300 ppm (IDLH).

## Section 9. Physical and Chemical Properties

<b>APPEARANCE AND ODOR:</b>	Colorless liquid with a pungent odor.
<b>ODOR THRESHOLD:</b>	2 - 5 ppm
<b>SOLUBILITY IN WATER:</b>	Miscible
<b>SPECIFIC GRAVITY OF VAPOR (air = 1):</b>	0.596 at 32 °F
<b>WEIGHT (per gallon):</b>	7.46 pounds to 7.99 pounds
<b>EVAPORATION RATE (water = 1):</b>	Similar
<b>PH:</b>	13+
<b>FORMULA:</b>	NH <sub>4</sub> OH (NH <sub>3</sub> + H <sub>2</sub> O)
<b>MOLECULAR WEIGHT:</b>	35.05 (NH <sub>4</sub> OH)
<b>VISCOSITY:</b>	1.7 40 °F (26% solution)
<b>PARTITION COEFFICIENT:</b>	Not applicable.
<b>DECOMPOSITION TEMPERATURE:</b>	Not applicable.
<b>FLAMMABILITY:</b>	
<b>FLASHPOINT:</b>	Not applicable.
<b>FLAMMABLE LIMITS OF AMMONIA VAPOR IN AIR:</b>	LEL/UEL 16% to 25% (listed in the <i>NIOSH Pocket Guide to Chemical Hazards</i> 15% to 28%).
<b>AUTO-IGNITION TEMPERATURE (ammonia vapors):</b>	1,204 °F (If catalyzed). 1,570 °F (If un-catalyzed).

### SOLUTION-SPECIFIC PHYSICAL DATA:

	20.5° Baume	25° Baume	26° Baume
<b>AMMONIA PERCENTAGE:</b>	18.5% to 19.5%	26.5% to 27.5%	29.4% to 30.0%
<b>WATER PERCENTAGE:</b>	81.5% to 80.5%	73.5% to 72.5%	70.6% to 70.0%
<b>SPECIFIC GRAVITY (water = 1):</b>	0.9309 to 0.9278 at 60 °F	0.9060 to 0.9030 at 60 °F	0.8974 to 0.8957 at 60 °F
<b>APPROXIMATE BOILING POINT:</b>	120 °F at 14.7 psia	88 °F at 14.7 psia	84.9 °F at 14.7 psia
<b>VAPOR PRESSURE:</b>	3.9 psia at 60 °F	6.9 psia at 60 °F	9.1 psia at 60 °F
<b>APPROXIMATE FREEZING POINT:</b>	-31 °F	-89 °F	-110 °F

## Section 10. Stability and Reactivity

### **REACTIVITY:**

Avoid ammonium hydroxide contact with chemicals such as mercury, chlorine, iodine, bromine, silver oxide or hypochlorites; they can form explosive compounds. Ammonia reacts with strong oxidizers, acids, halogens (including chlorine bleach), and salts of silver, zinc, copper, and other heavy metals.

### **CHEMICAL STABILITY:**

Stable under normal ambient conditions of temperature and pressure.  
Will not polymerize.

### **POSSIBILITY OF HAZARDOUS REACTIONS:**

Ammonium hydroxide will react exothermically with acids.  
Ammonia vapors are released when heated.

### **CONDITIONS TO AVOID:**

Avoid ammonium hydroxide contact with chlorine, which forms a chloramine gas, which is a primary skin irritant and sensitizer.

### **INCOMPATIBLE MATERIALS:**

Ammonium hydroxide has a corrosive reaction with galvanized surfaces, copper, brass, bronze, aluminum alloys, mercury, gold and silver.

### **HAZARDOUS DECOMPOSITION PRODUCTS:**

Ammonia will be liberated if heated. Hydrogen will be released on heating ammonia above 450 °C (842 °F).

## Section 11. Toxicological Information

**Potential health effects:** Ammonia is an irritant and corrosive to the skin, eyes, respiratory tract and mucous membranes. May cause severe chemical burns to the eyes, lungs and skin. Skin and respiratory related diseases could be aggravated by exposure. The extent of injury produced by exposure to ammonia depends on the duration of the exposure, the concentration of the liquid or vapor and the depth of inhalation.

**Exposure Routes:** Inhalation (vapors), skin and/or eye contact (vapors, liquid), ingestion (liquid).

### **Symptoms of acute exposure:**

**Inhalation:** Acute exposure to vapor may result in severe irritation of the respiratory tract. May cause dyspnea (breathing difficulty), wheezing, chest pain, bronchospasm, pink frothy sputum, pulmonary edema or respiratory arrest. Respiratory injury may appear as a delayed phenomenon. Pulmonary edema may follow chemical bronchitis.

**Eyes:** Vapors may cause irritation. Effects of direct contact may range from irritation and lacrimation (tearing) to severe corrosive injury and blindness.

**Skin:** Irritation, corrosive burns, blister formation (vesiculation) may result. Contact with liquid may produce caustic burns.

**Ingestion:** May cause corrosion to the mouth, throat, esophagus and stomach with perforation and peritonitis. Extreme exposure may result in death from spasm, inflammation or edema.

**Chronic Exposure:** Repeated exposure to ammonia may cause chronic irritation of the eyes and respiratory tract.

### **Toxicity:**

LC<sub>50</sub> - 5131 mg/m<sup>3</sup> (7338 ppm) to 11,592 mg/m<sup>3</sup> (16,600 ppm), 60 minute exposure, Rat.  
LD<sub>50</sub> - 350 mg / kg (Oral / Rat).

Not listed in the National Toxicology Program (NTP).

Not recognized by OSHA as a carcinogen.

Not listed as a carcinogen by the International Agency for Research on Cancer (IARC monograph).

Germ cell mutagenicity information is not available. Reproductive toxicity information is not available.

## Section 12. Ecological Information

Ammonia is harmful to aquatic life at very low concentrations. Notify local health and wildlife officials and operators of any nearby water intakes upon contamination of surface water.

### Toxicity:

Terrestrial plants: LOEC = 3-250 ppm NH<sub>3</sub>.

Aquatic plants: LOEC = 0.5-500 mg NH<sub>3</sub>-N/L.

Acute toxicity to invertebrates: 48 h LC50 = 2.94 mg un-ionized NH<sub>3</sub>-N/L.

Chronic toxicity to invertebrates: NOEC = 0.163- 0.42 mg un-ionized NH<sub>3</sub>/L.

Acute toxicity to fish: 96-h: LC50= 0.09 – 3.51 mg un-ionized NH<sub>3</sub>/L.

Chronic toxicity to fish: NOEC=0.025-1.2 mg un-ionized NH<sub>3</sub>/L.

**Environmental Fate Information:** Ammonia dissipates relatively quickly in ambient air and rapidly returns to the soil via combination with sulfate ions or washout by rainfall. Ammonia strongly adsorbs to soil, sediment particles and colloids in water under aerobic conditions. Biodegradation of ammonia to nitrate occurs in water under aerobic conditions resulting in a biological oxygen demand (BOD).

### Persistence/Degradability:

Biodegradable in soil. Ozonation in the air. Soluble in water.

### Bioaccumulative Potential:

Not applicable.

### Mobility in Soil:

No additional information available.

### Other Adverse Effects:

No additional information available.

## Section 13. Disposal Considerations

Dispose of unused contents/container in accordance with local/regional/national regulations as applicable.

Listed as hazardous substance under Clean Water Act (CWA) (40 CFR 116.4 and 40 CFR 117.3).

Classified as hazardous waste under Resource Conservation and Recovery Act (RCRA) (40 CFR 261.22 Corrosive #D002) if disposed of in original form.

Suitably diluted product may be utilized as fertilizer on agricultural land.

For hazardous waste regulations information call the RCRA Hotline (800) 424-9346, or visit the US EPA website.

## Section 14. Transport Information

### US Department of Transportation

HAZARD CLASS: 8 (Corrosive Material)

PROPER SHIPPING DESCRIPTION: UN2672, Ammonia Solutions, 8, PG III, RQ

PLACARD: Corrosive



IDENTIFICATION NUMBER: UN 2672

### ENVIRONMENTAL HAZARDS:

IMDG, Known Marine Pollutant: No

United Nations Model Regulations, Environmentally Hazardous: No

## Section 15. Regulatory Information

The material is subject to the reporting requirements of Section 304, Section 312 and Section 313, Title III of the Superfund Amendments and Reauthorization Act (SARA) of 1986 and 40 CFR 372.

Under Section 313, as of June 30, 1995, this material is reportable with the following qualifications: 10% of total aqueous ammonia is reportable as Ammonia (CAS #: 7664-41-7) under this listing.

Under the Comprehensive Environmental Response, Compensation, and Liability Act of 1980 (CERCLA), Section 103, any environmental release of this chemical equal to or over the reportable quantity of 1,000 pounds (as NH<sub>4</sub>OH) must be reported promptly to the National Response Center, Washington, D.C. (1-800-424-8802).

Toxic Substances Control Act (TSCA): This material and its components are listed in the TSCA Inventory.

EPA Hazard Categories – Immediate: Yes; Delayed: No; Fire: No; Sudden Release: No; Reactive: No

Clean Air Act – Section 112(r): Material is listed under EPA's Risk Management Program (RMP), 40 CFR Part 68 at concentrations greater than 20% and storage/process amounts greater than the Threshold Quantity (TQ) of 20,000 pounds of contained ammonia (CAS #: 7664-41-7).

The chemical is listed under Department of Homeland Security regulation 6 CFR Part 27, Chemical Facility Anti-Terrorism Standards at storage / process amounts greater than the threshold quantity of 20,000 pounds (ammonia concentration 20% or greater).

OSHA (Occupational Safety & Health Administration): This material is considered to be hazardous as defined by the OSHA Hazard Communication Standard 29 CFR 1910.1200. This material is subject to Process Safety Management requirements of 29 CFR 1910.119 if maintained on-site, including storage and process, in quantities of 15,000 pounds or greater (>44% ammonia by weight).

## Section 16. Other Information

Preparation Information: Revision Date May 1, 2015  
Replaces all previously dated versions.

Prepared by: HJS

Revisions to this Safety Data Sheet have been created to comply with the requirements of the OSHA Hazard Communication Final Rule issued in 2012 (HazCom 2012).

### Acronyms:

ACGIH: American Conference of Governmental Industrial Hygienists  
ANSI: American National Standards Institute  
CAS: Chemical Abstracts Service  
CFR: Code of Federal Regulations  
DHS: Department of Homeland Security  
DOT: Department of Transportation  
EPA: Environmental Protection Agency  
HMIS: Hazardous Materials Identification System  
IARC: International Agency for Research on Cancer  
IDLH: Immediately Dangerous to Life or Health  
IMDG: International Maritime Dangerous Goods  
NFPA: National Fire Protection Association  
NIOSH: National Institute for Occupational Safety and Health  
NTP: National Toxicology Program  
OSHA: Occupational Safety and Health Administration  
PEL: Permissible Exposure Limit  
PPM: Parts Per Million  
RCRA: Resource Conservation and Recovery Act  
REL: Recommended Exposure Limit  
SCBA: Self Contained Breathing Apparatus

STEL: Short Term Exposure Limit  
TLV: Threshold Limit Value  
TWA: Time Weighted Average

Disclaimer:

The information, data, and recommendations in this safety data sheet relate only to the specific material designated herein and do not relate to use in combination with any other material or in any process. To the best of our knowledge, the information, data, and recommendations set forth herein are believed to be accurate. We make no warranties, either expressed or implied, with respect thereto and assume no liability in connection with any use of such information, data, and recommendations. Judgements as to the suitability of the information contained herein for the party's own use or purposes are solely the responsibility of that party. Any party handling, transferring, transporting, storing, applying or otherwise using this product should review thoroughly all applicable laws, rules, regulations, standards and good engineering practices. Such thorough review should occur before the party handles, transfers, transports, stores, applies or otherwise uses this product.